



UNIVERSITY OF CHEMISTRY AND TECHNOLOGY, PRAGUE

Faculty of Food and Biochemical Technology

Department of Food Analysis and Nutrition

Overview of advanced instrumental techniques employed in food and feed analysis

Vit Kosek

Food composition

Natural components

Natural toxins

Antinutrition
comps.

Primary
sensorically
active comp.

Antioxidants and
other biologically
active
components

Nutrients

proteins
lipids
saccharides
minerals
vitamins

Fiber



Contaminants

Environmental
contaminants

Pesticide / veterinary
drug residues

Migrants from plastics

Toxic metals

Processing products

Additives

Biotechnology products

Food and feed analysis

Applications:

- Regulatory
- Food safety
- Quality control
- Research and development



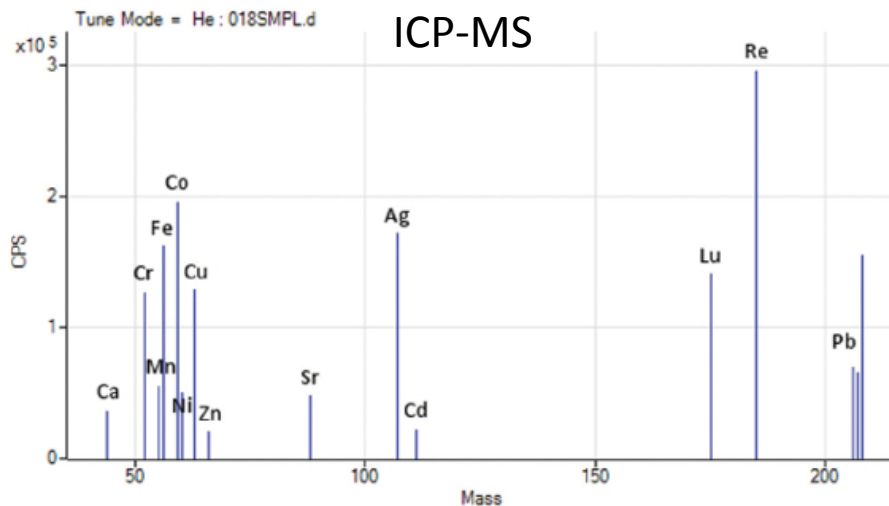
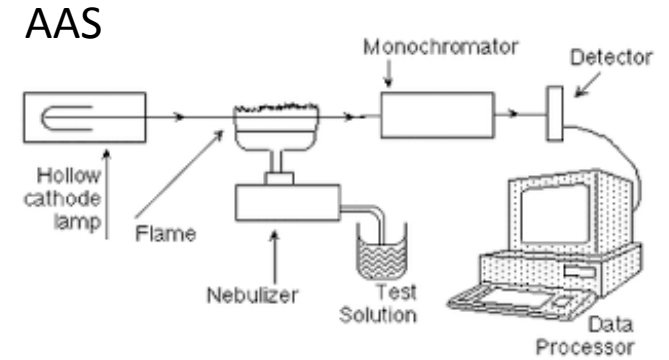
We need the methods to be...

- Precise
- Reproducible
- Accurate
- Simple
- Cheap
- Fast
- Sensitive
- Specific
- Safe
- Destructive/Non-destructive
- Online/Offline
- Official

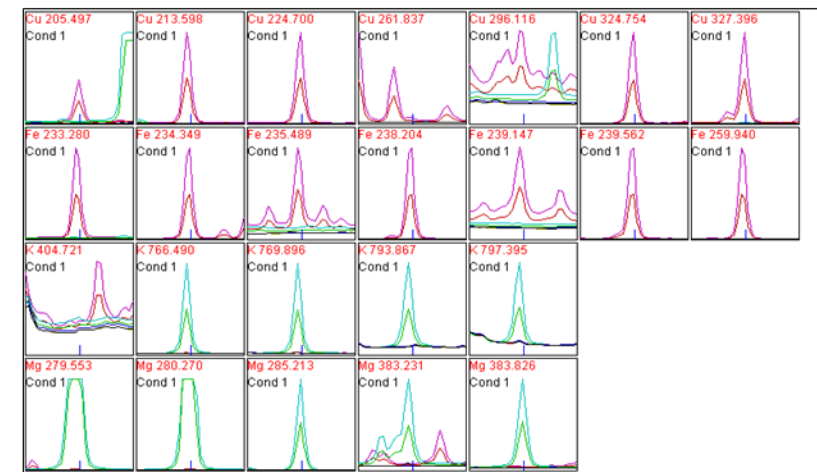
Techniques must be always fit for purpose!

Elemental analysis

- Atomic absorption spectroscopy
- ICP-atomic emission spectroscopy
- ICP- mass spectrometry

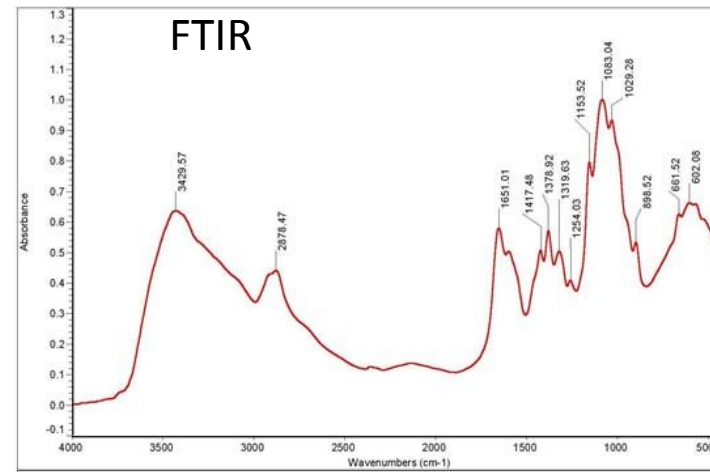
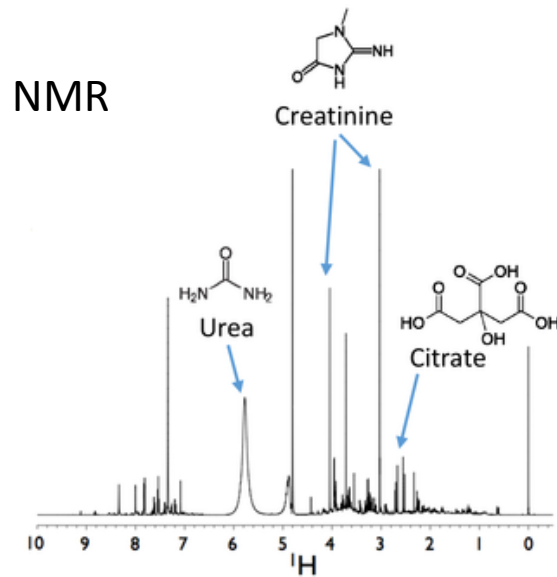
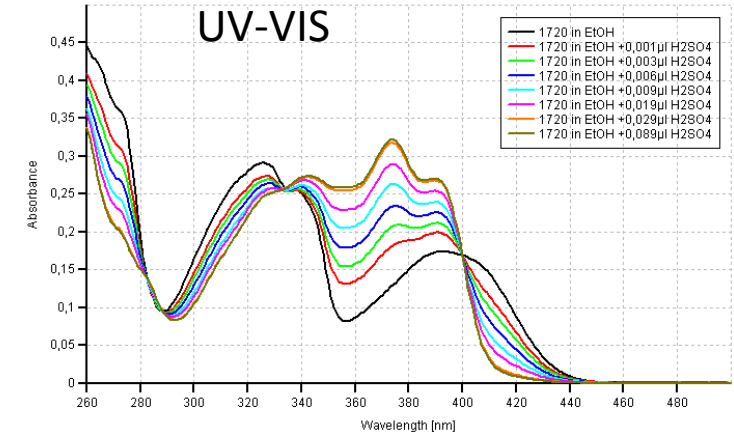


ICP-AES



Molecular spectroscopy

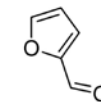
- UV-VIS
- Infra-Red
- Nuclear Magnetic Resonance



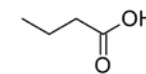
Gas chromatography

- Separation of sample constituents in gas phase
- On the basis of volatility and structure
- Analytes need to be sufficiently volatile and thermally stable
- Analytes usually up to 1000 Da

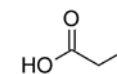
More on this topic: **Michal Stupák**



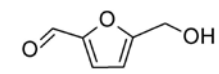
Furfural



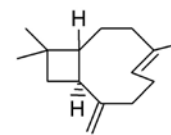
Butanoic acid



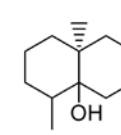
Propanoic acid



5-Hydroxy-methyl-furfural



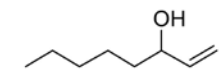
β -caryophyllene



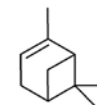
Geosmin



2-Methyl isoborneol



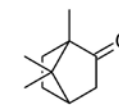
1-octen-3-ol



α -pinene



Camphene



Camphor



Methanol



Acetaldehyde

Liquid chromatography

- Separation of sample constituents in liquid state
- Wide range of analytes are separable
- No need for temperature stability
- Several mechanisms:
 - Hydrophobic interactions (reverse phase)
 - Polar interactions and hydrogen bonds (normal phase, HILIC)
 - Charge interactions (ion Exchange)

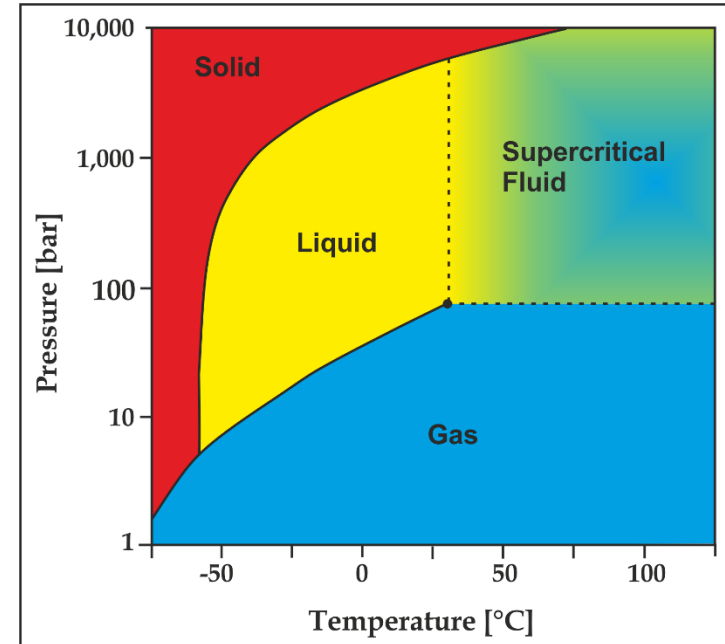
More on this topic: **Vojtěch Hrbek**

Supercritical fluid chromatography

- Mobile phase - **supercritical CO₂** ($T_{crit} = 31\text{ °C}$, $P_{krit} = 7390\text{ kPa}$)
- **Fluid with low viscosity and high diffusivity** → high separation efficiency, shortened time of analysis
- **Polarity of supercritical CO₂ ~ hexane**
- Amenable for analytes with wide range of polarities

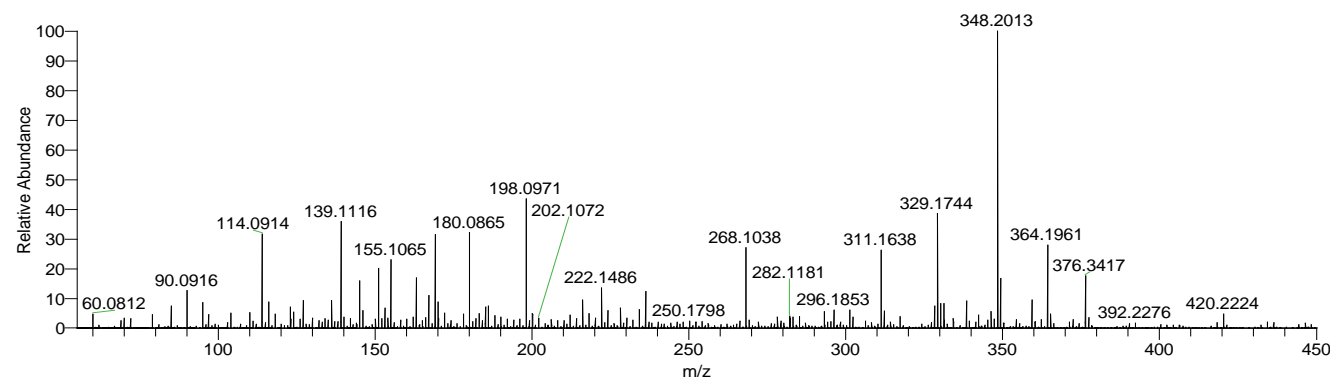
More on this topic with:

Beverly Bělková, Michaela Rektorisová

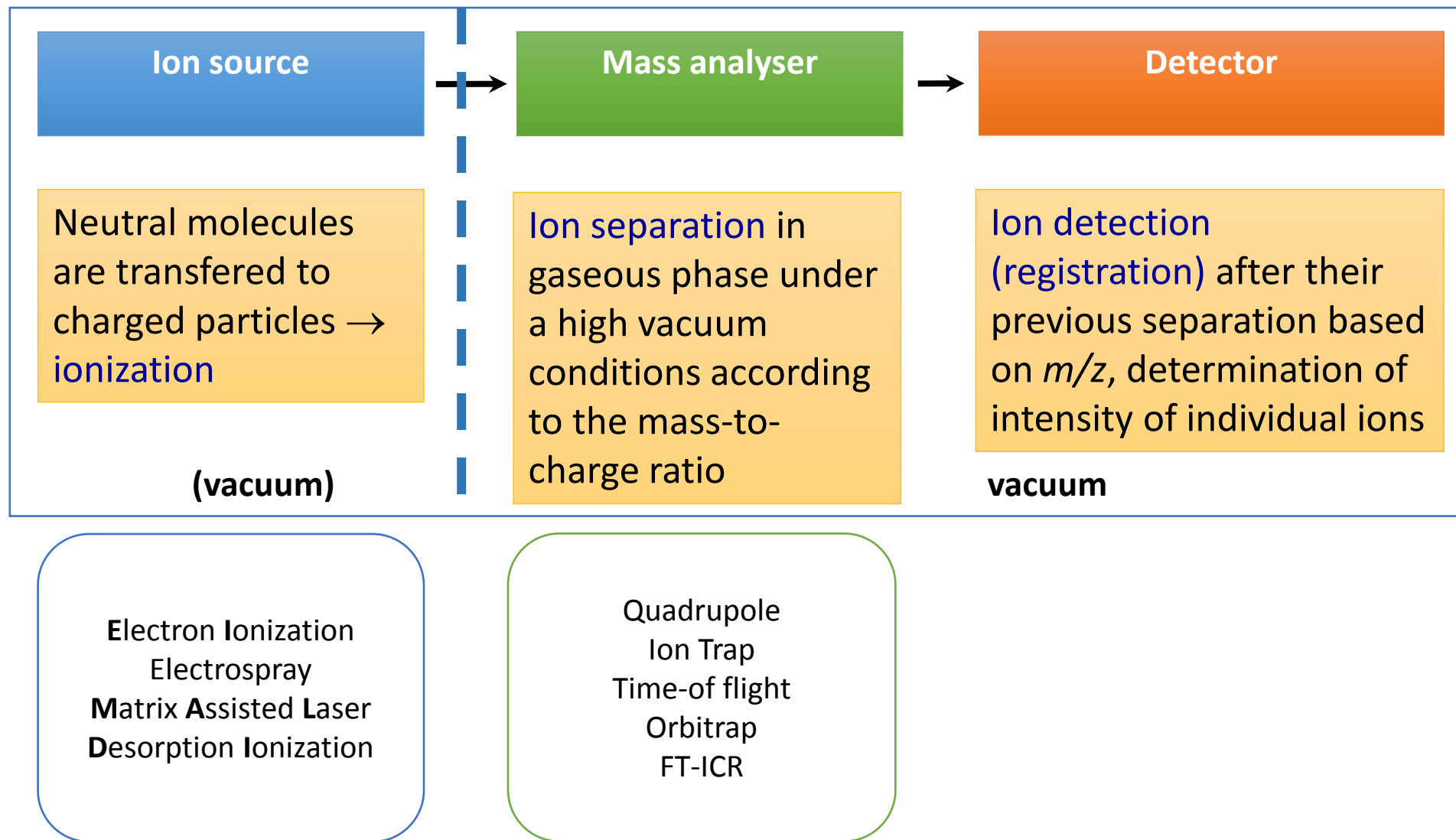


Mass spectrometry

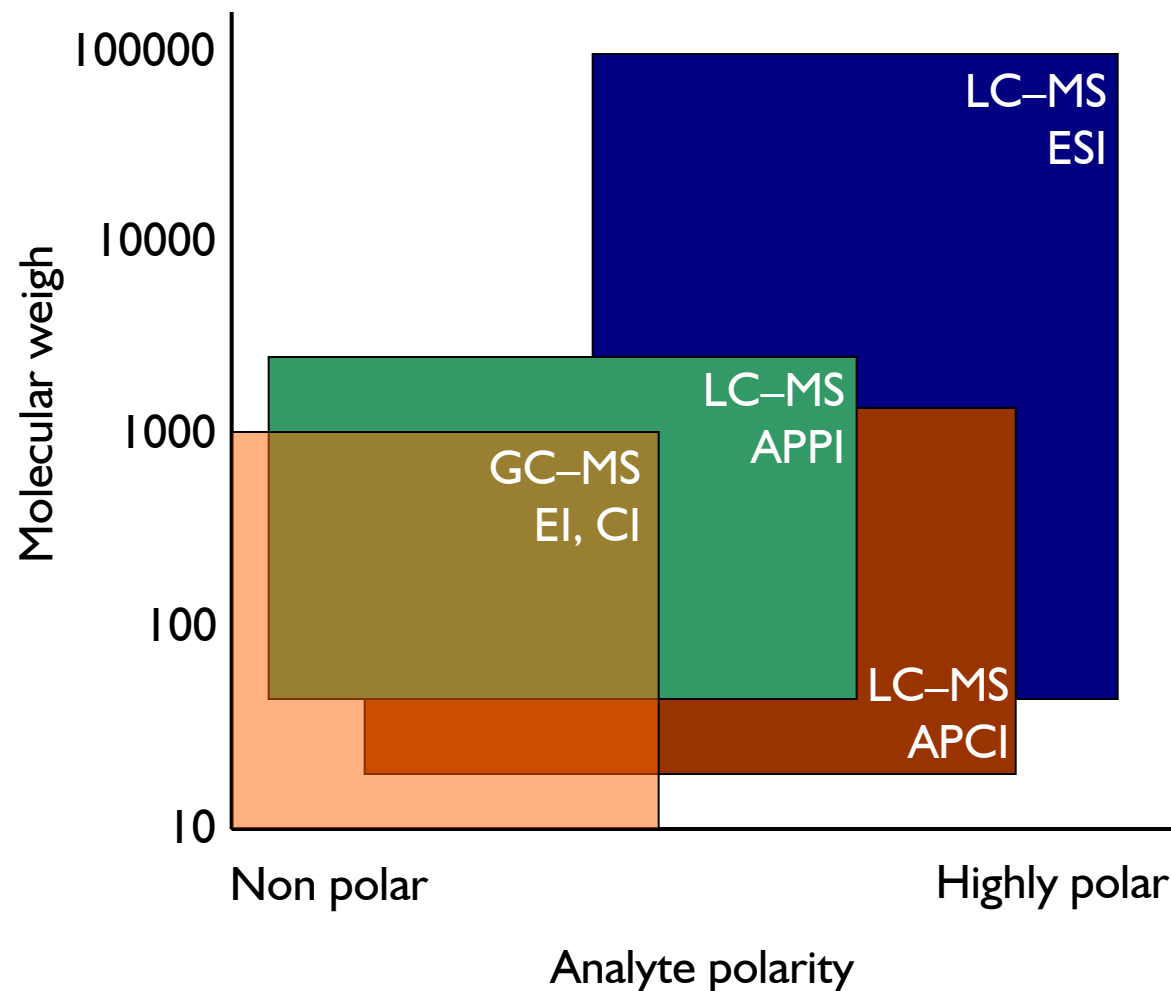
- Weighing molecules
- Molecules need to be ionised
- Ions can be manipulated with in electric or magnetic field
- Mass spectrum: m/z X intensity
- Destructive X very sensitive
- Specific



PARTS OF MASS SPECTROMETER

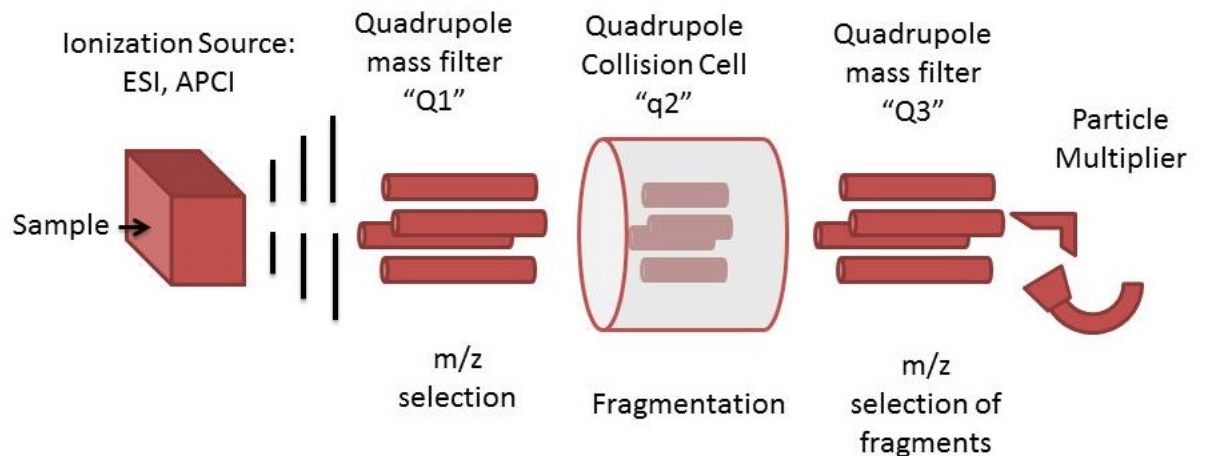


Mass spectrometry and separation techniques



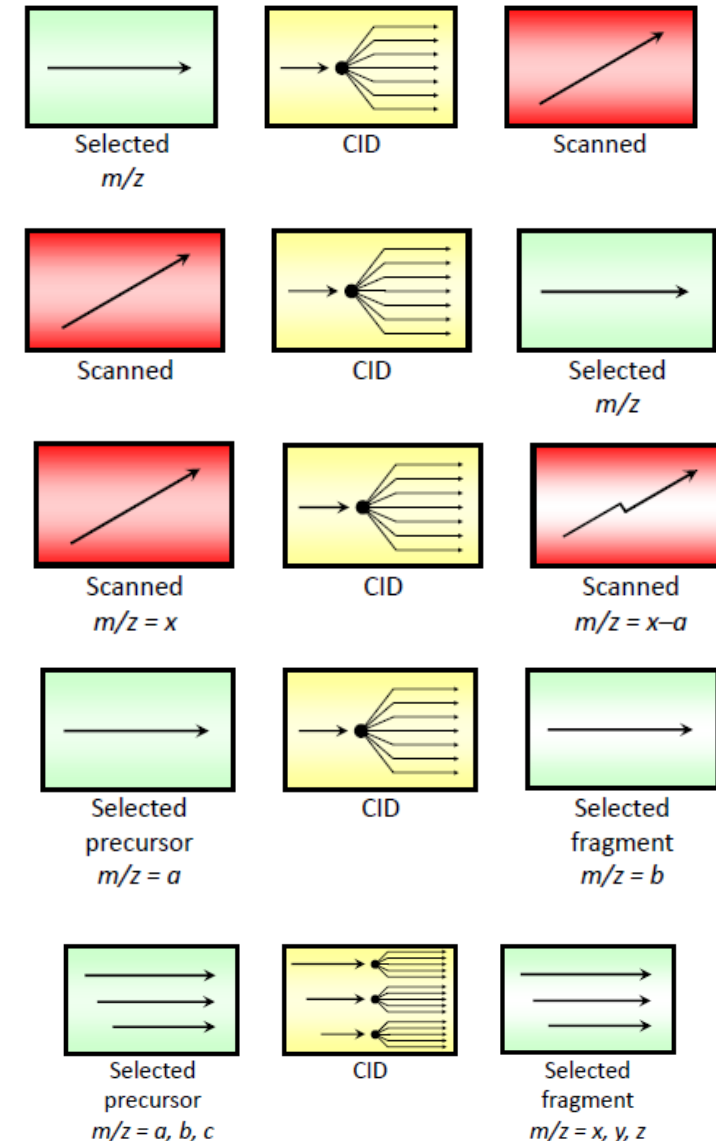
TANDEM MS

- A method comprising at **least two levels of mass analysis** steps: either in connection with a dissociation process or a chemical reaction that causes a change in the ion mass or ion charge
- MS/MS methods involve the **activation of the selected ion** (precursor)
- Activation of ions in space or in time



MODES OF TANDEM MS

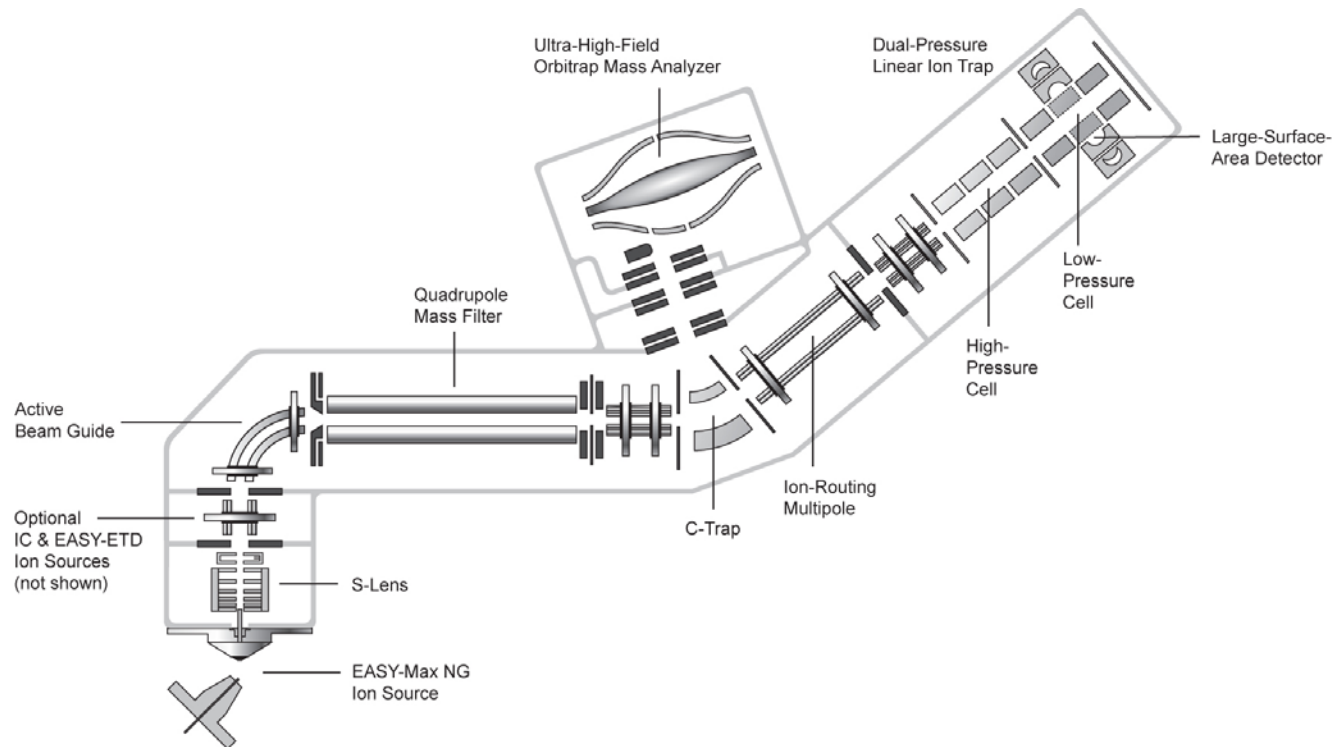
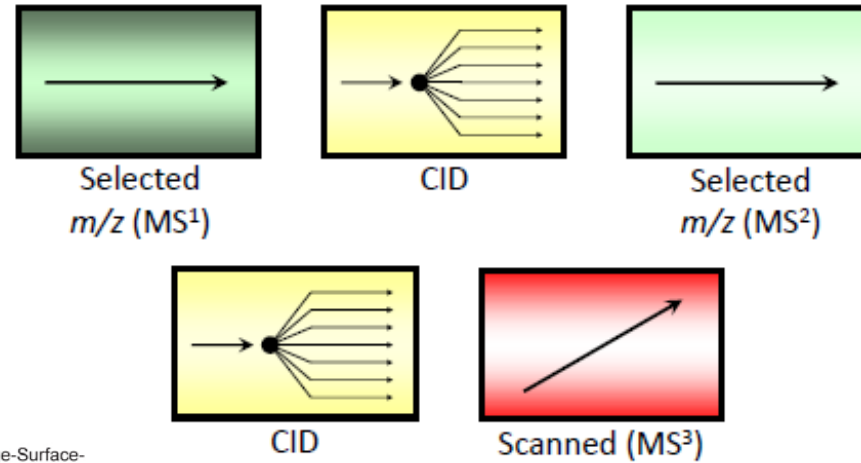
- Product ion scan
- Precursor ion scan
- Neutral loss scan
- Selected reaction monitoring
- Multiple reaction monitoring



MODES OF TANDEM MS

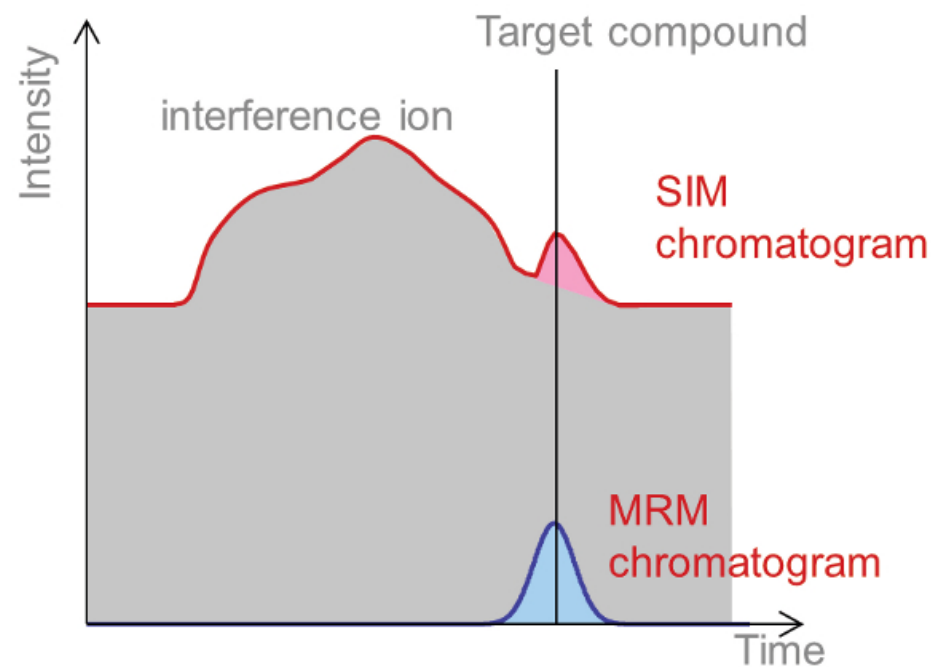
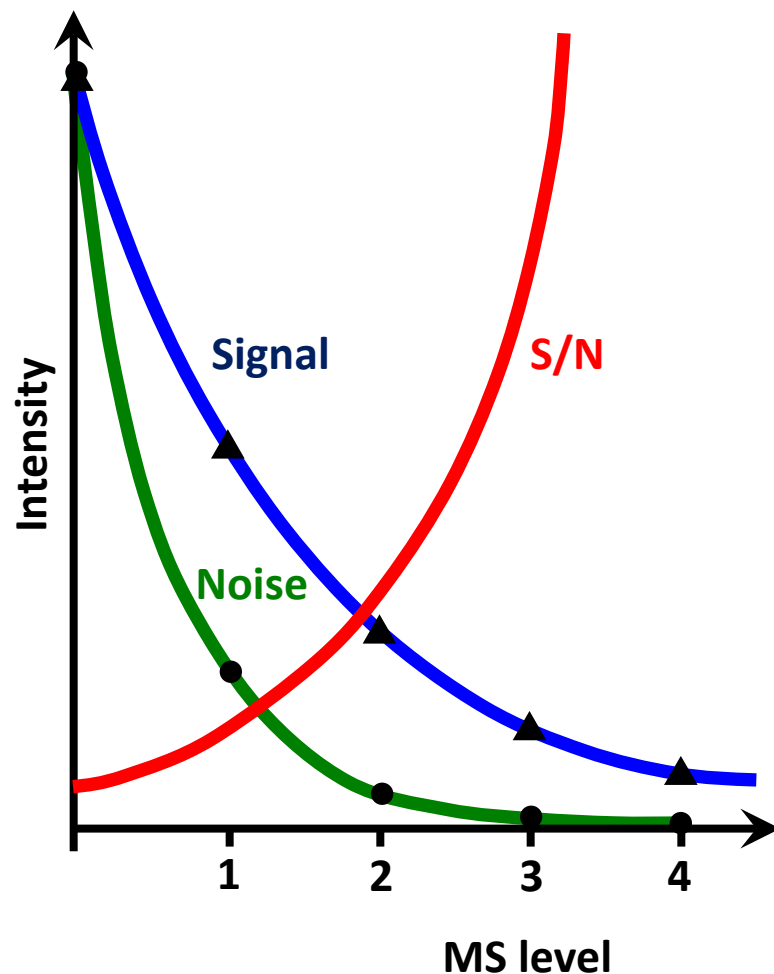
- Scan MS^n

- *Applicable for ion traps*



Effect of tandem MS

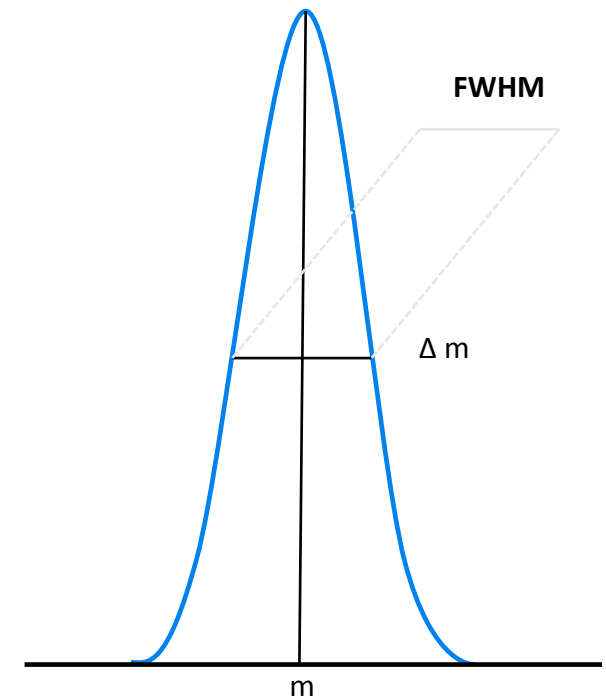
- Scan MSⁿ: selectivity × sensitivity



source: R.G. Cooks, K.L. Busch, J. Chem. Educ.
59(11) (1982) 926–933

High resolution mass spectrometry

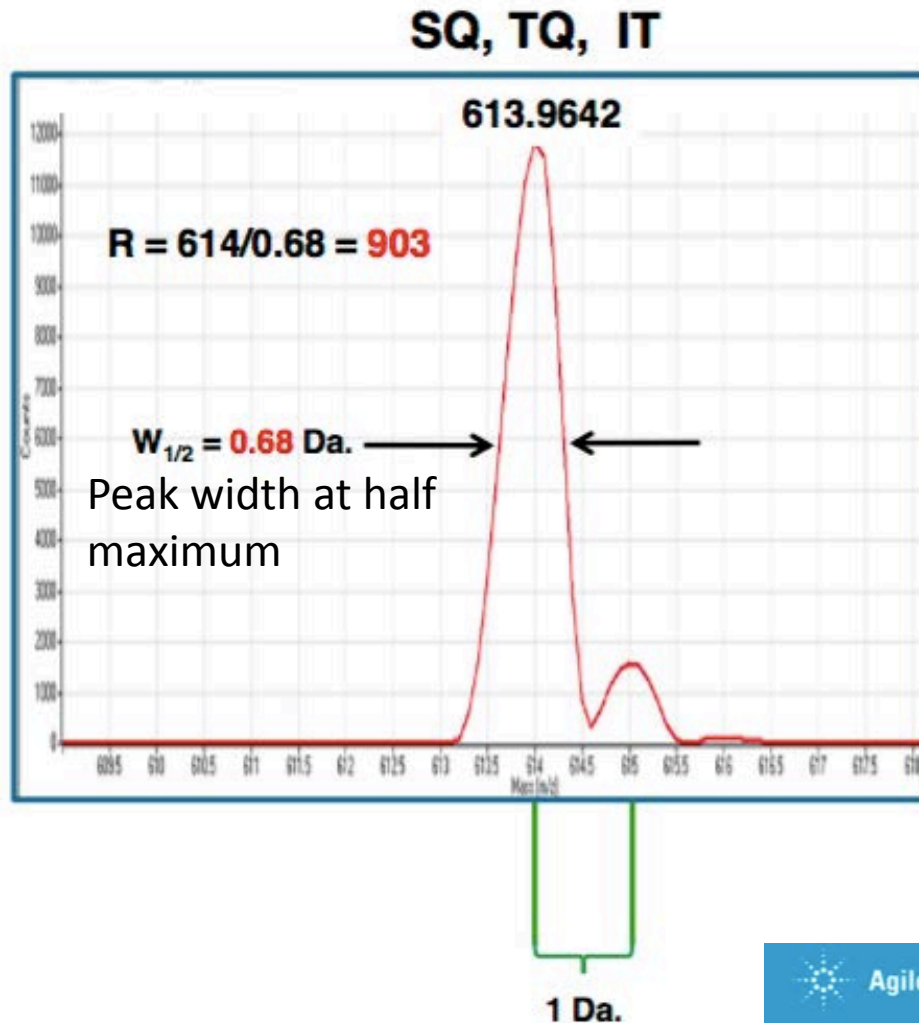
- Mass spectrometer has **high resolving power**
- Definition according to the width of one peak
- *Full Width at Half Maximum* (FWHM)
 - Mass difference expressed as the peak width of a given mass peak measured (in mass units) at 50% of its height



$$RP = m / \Delta m$$

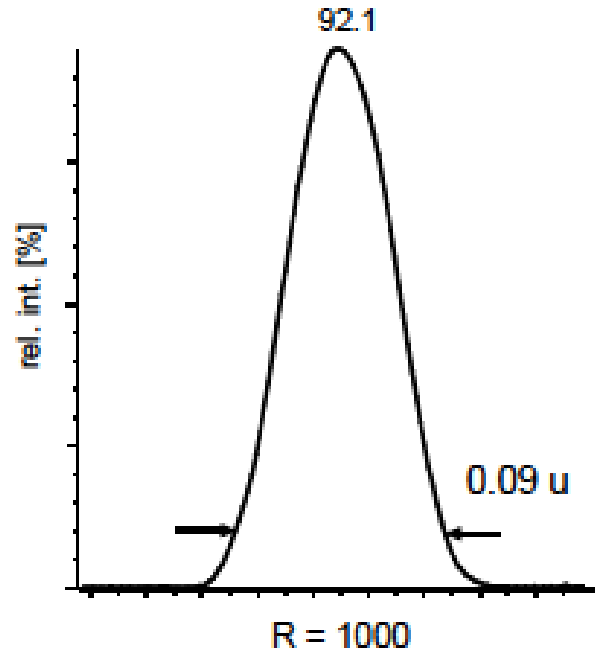
High resolution mass spectrometry

$m/z = 613.964203$



High resolution mass spectrometry

- A mixture of xylene (m/z 92.0581) and toluene (m/z 92.0626) at different settings of resolution



High resolution mass spectrometry

Mass accuracy:

- The deviation between measured mass (accurate mass) and calculated mass (exact mass) of an ion expressed as an error value (mDa, ppm)
- Important for structural interpretation (calculation of elemental composition)

$$\Delta \text{ (ppm)} = \frac{m_{\text{exp.}} - m_{\text{teor.}}}{m_{\text{teor.}}} \cdot 10^6$$

$$\Delta \text{ (mDa)} = (m_{\text{exp.}} - m_{\text{teor.}}) \cdot 10^3$$

FT-ICR

- RP: up to 10,000 k
- MA: below 1 ppm
- COST: +++++

ORBITRAP

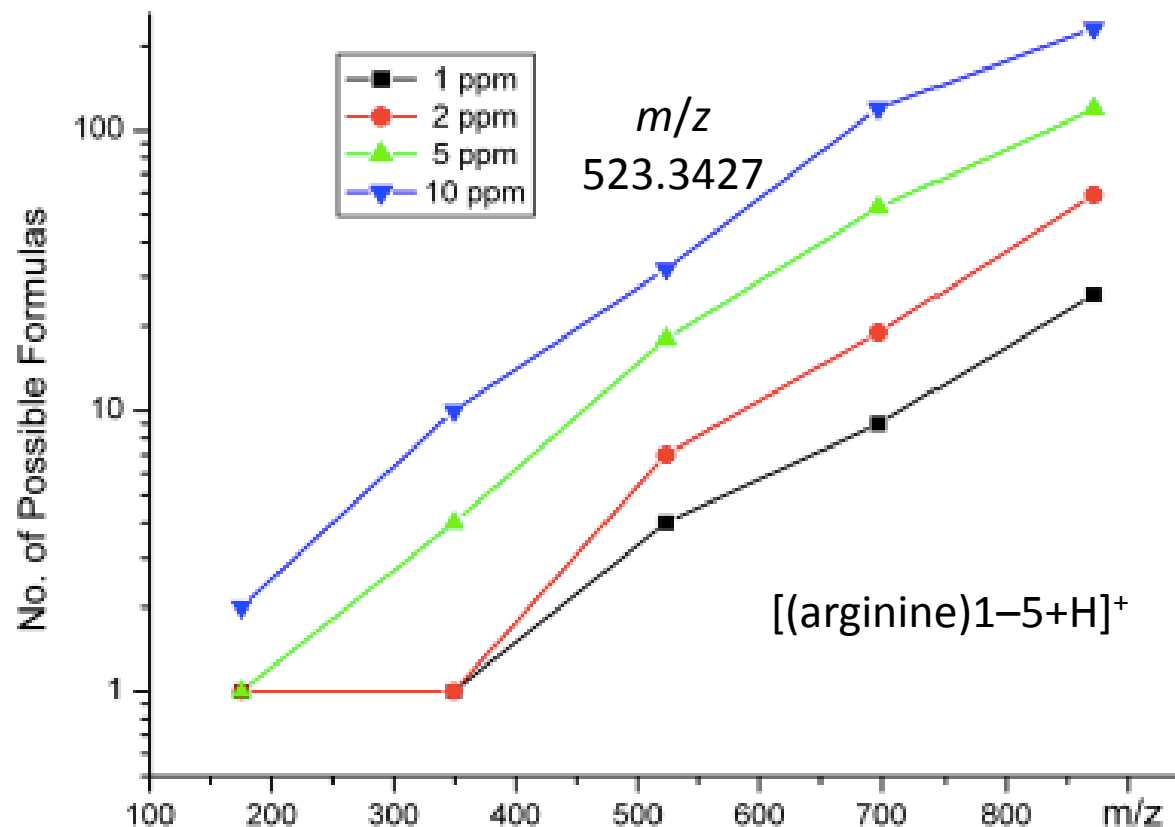
- RP: up to 450 k
- MA: <1 – 3 ppm
- COST: ++++

TIME-OF-FLIGHT

- RP: up to 50 k
- MA: <1 – 5 ppm
- COST: +++(+)

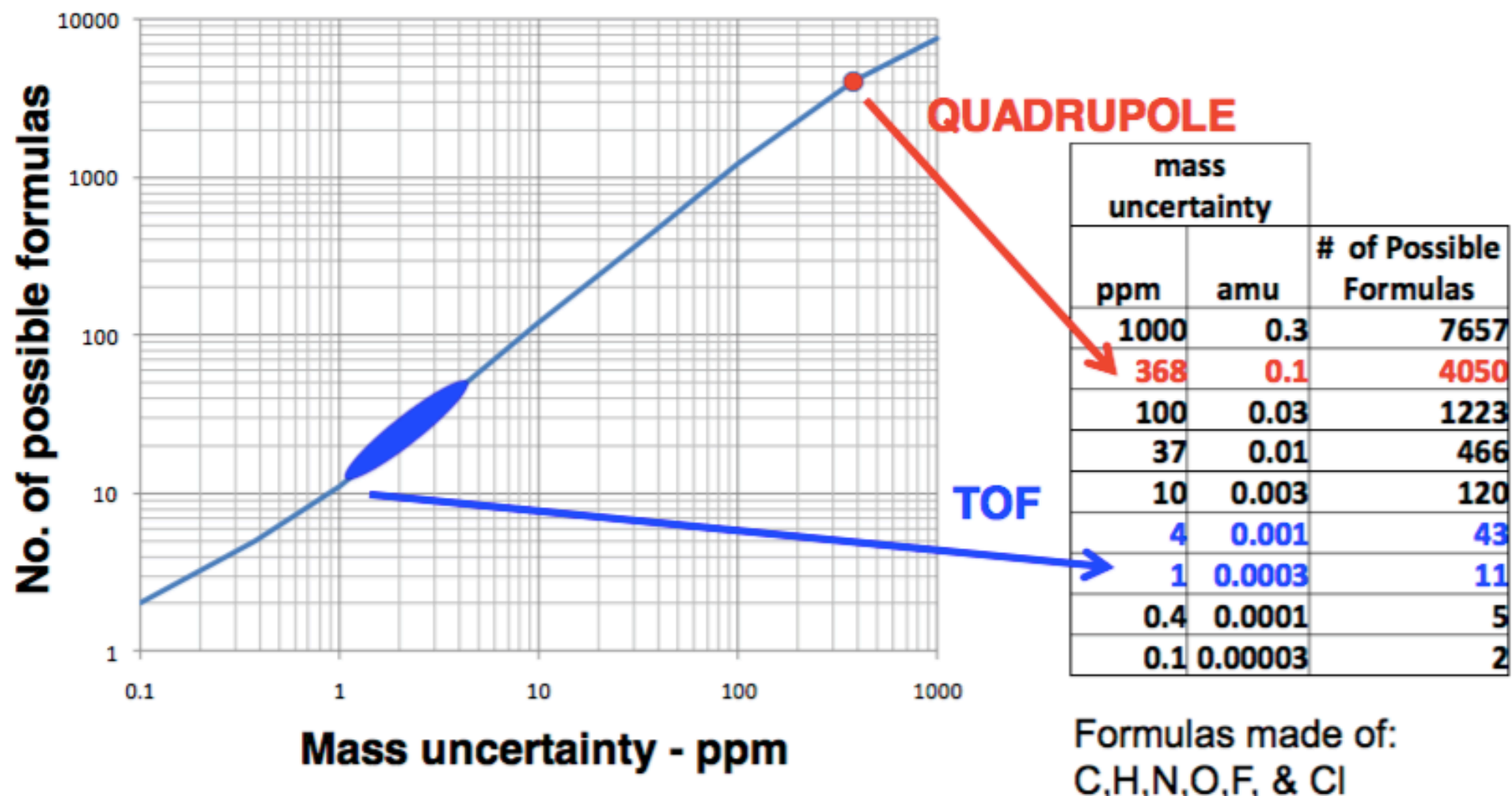
Why do we need accuracy and precision?

Molecular formulas based on a free selection among the elements C, H, N, O as a function of relative mass error vs. m/z .



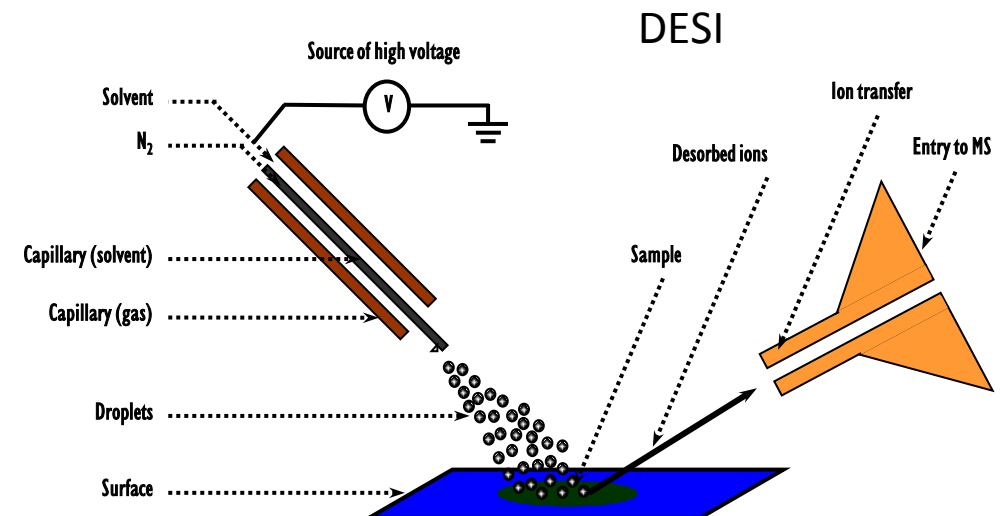
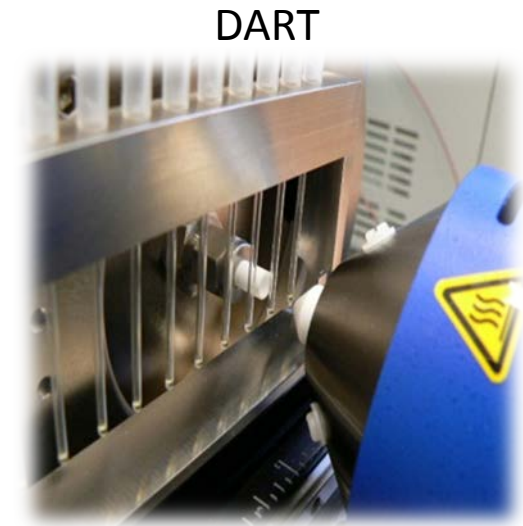
The higher the mass error the larger number of candidates

Possible chemical formulas for $m/z = \text{C}_{10}\text{F}_8 = 271.98667$



Ambient mass spectrometry

- Sample ionization at atmospheric pressure
- Usually no separation
- Fast response
- MS imaging

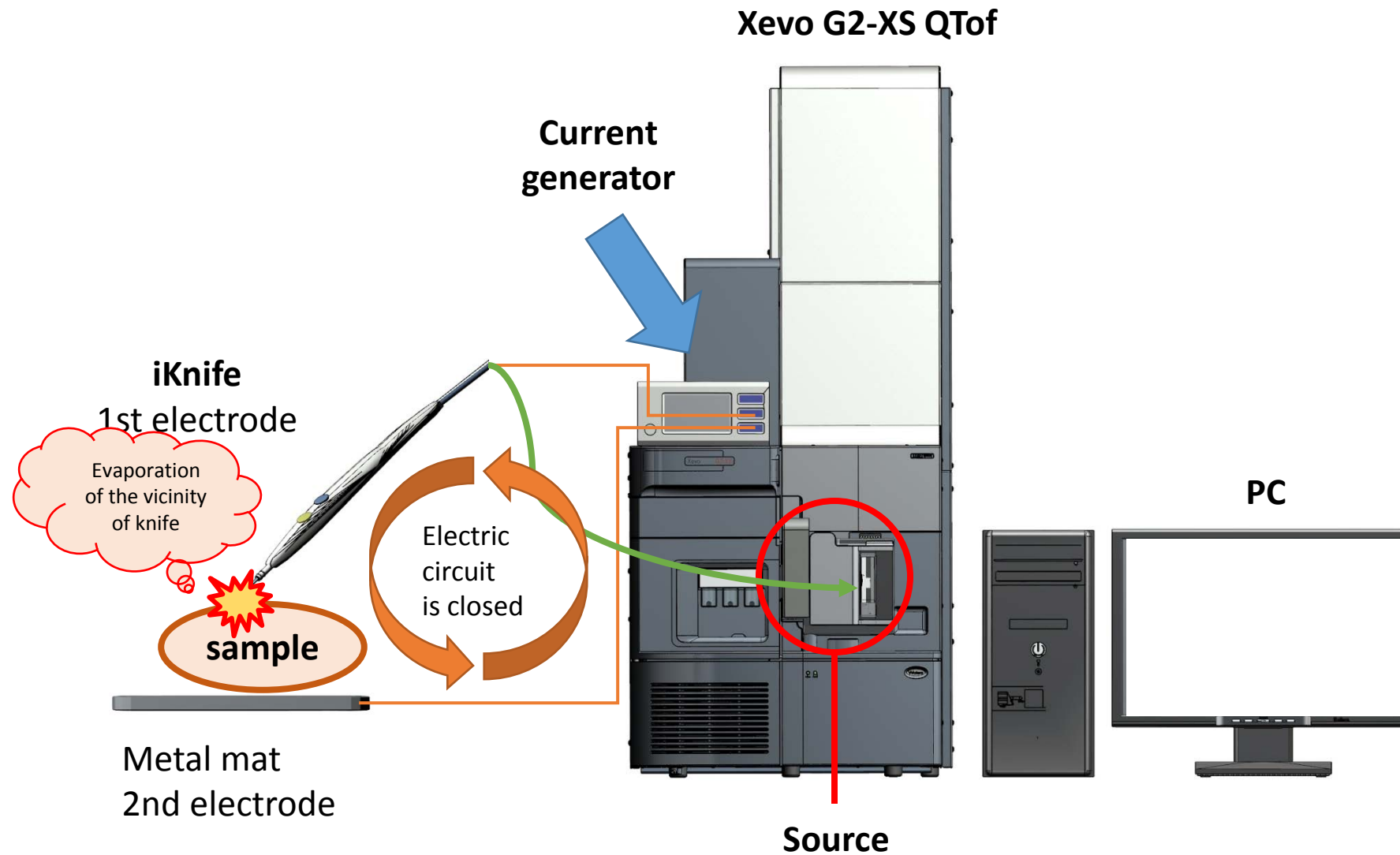


REIMS

- **R**apid **E**vaporative **I**onization **M**ass **S**pectrometry
- First electrosurgical knife **1926**
- The hyphenation of electrosurgical knife and mass spec in **2010**
- Developed for cancer surgery

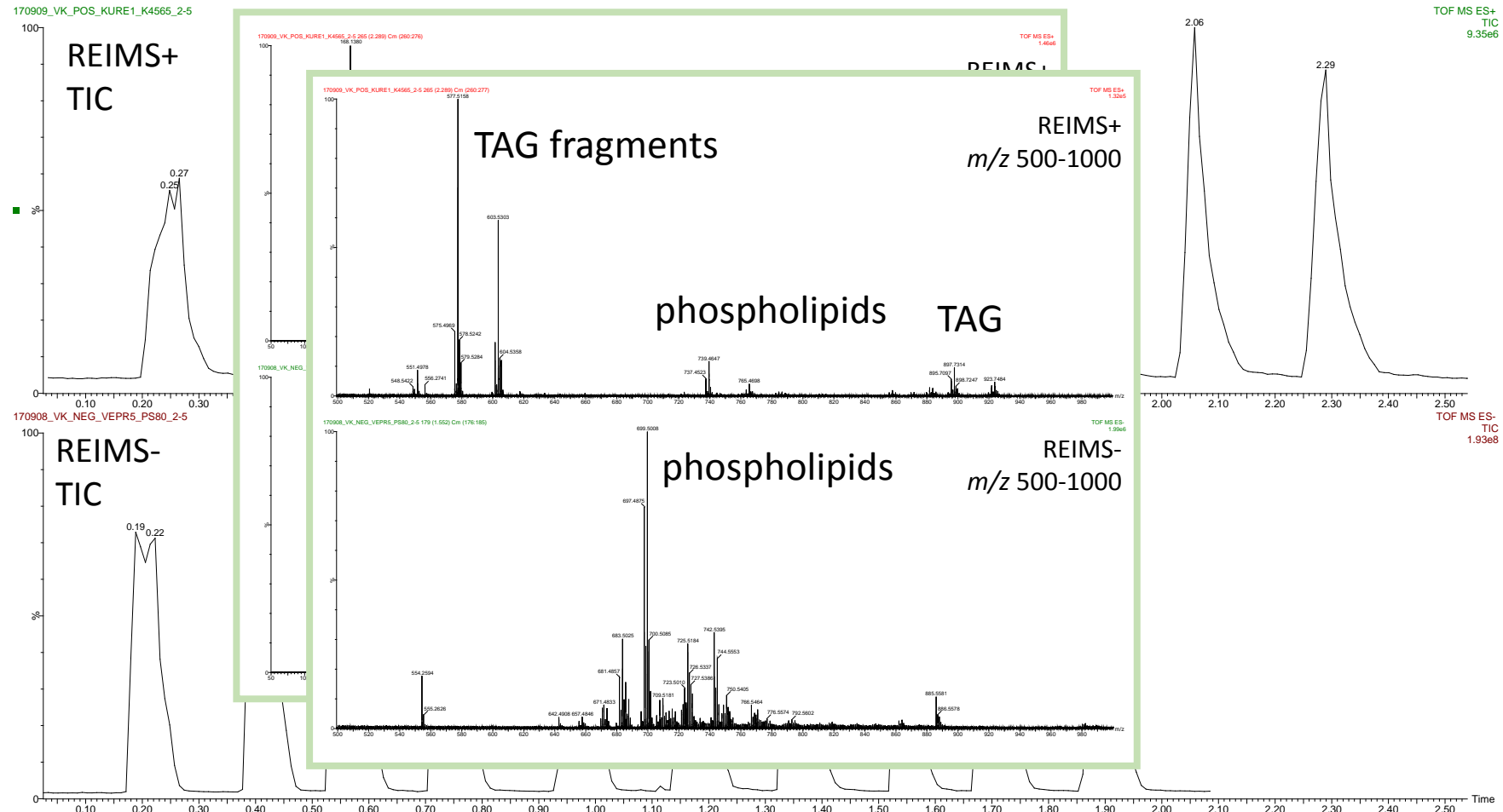


REIMS system



REIMS data

Chronograms

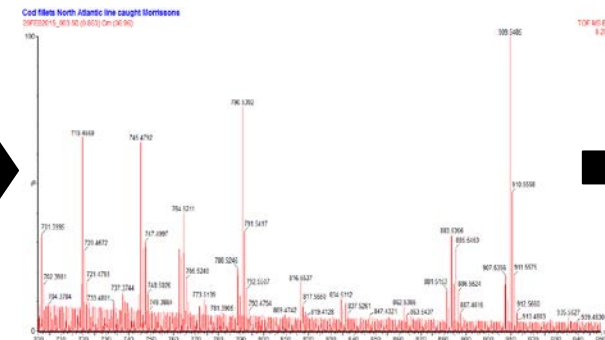


REIMS method workflow

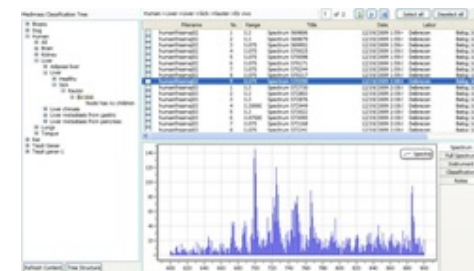
iKnife
+
Mass spec



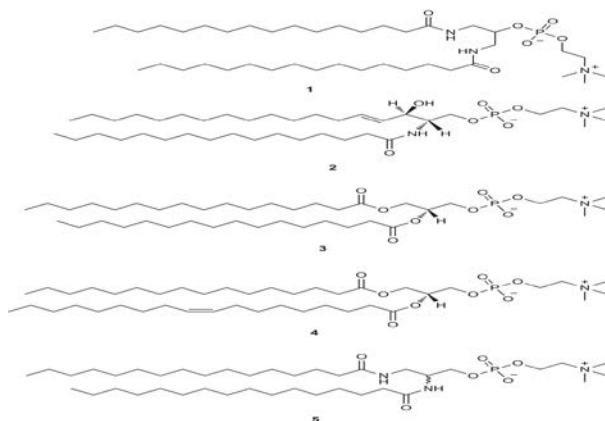
spectra



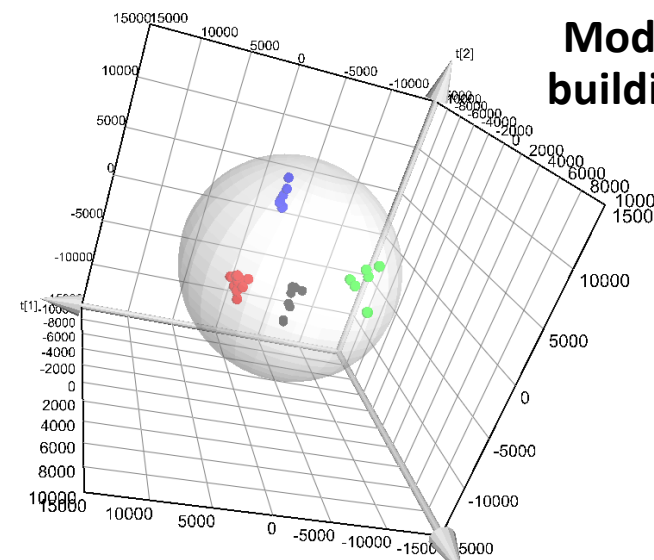
Database
creation



Tentative identification of important variables



Scores Comp[1] vs. Comp[3] vs. Comp[2], colored by Condition



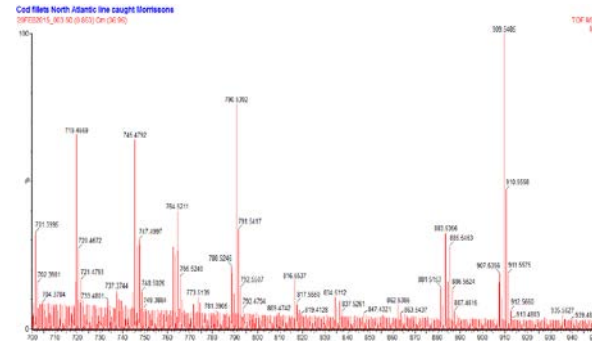
Model
building

MVA

■ Bovine muscle
■ Chicken muscle
■ Ovine muscle
■ Porcine muscle
● Ellipse: Hotelling T2 (0.95)

Authentication by REIMS

iKnife
+
Mass spec



Why REIMS?

Advantages:

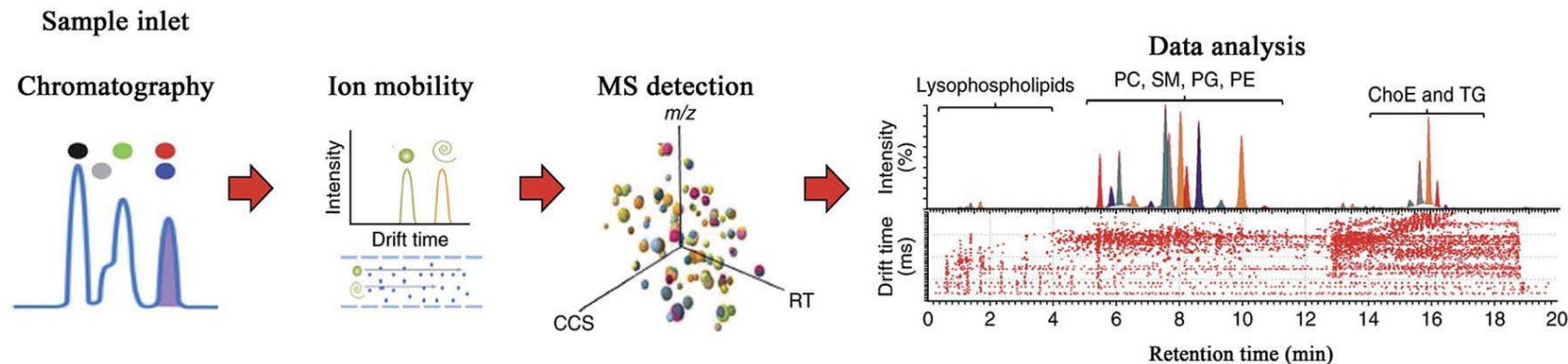
- Quick alternative PCR, MS a LC-MS methods
- Possibility of mobile instruments

Disadvantages:

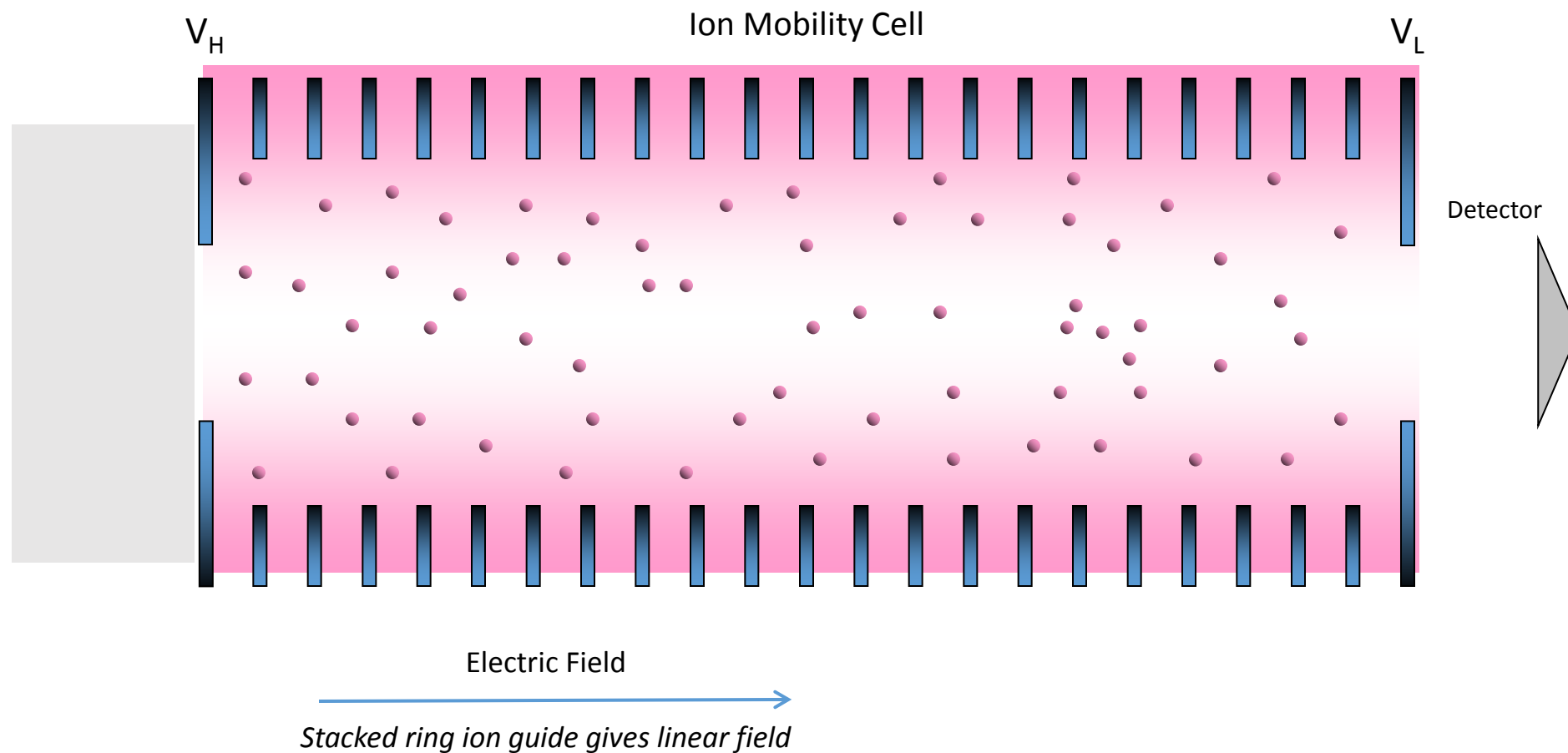
- Low sensitivity
- Limited number of matrices

Ion mobility MS

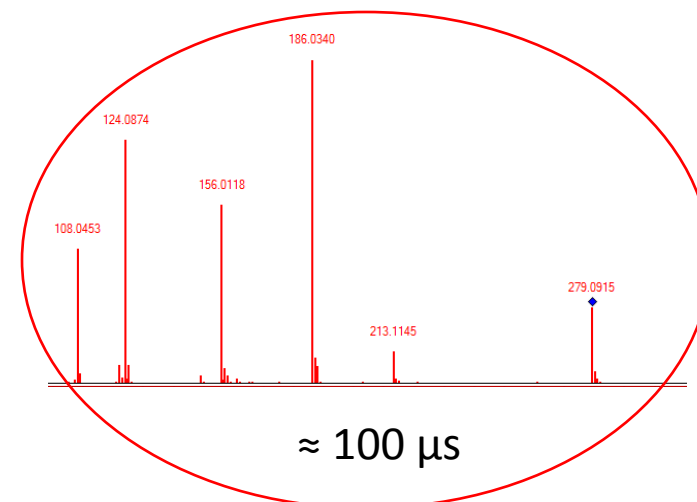
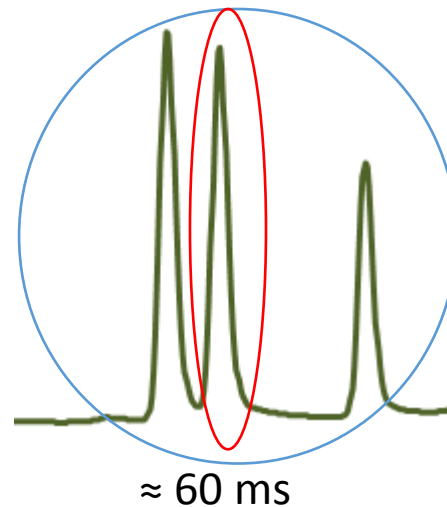
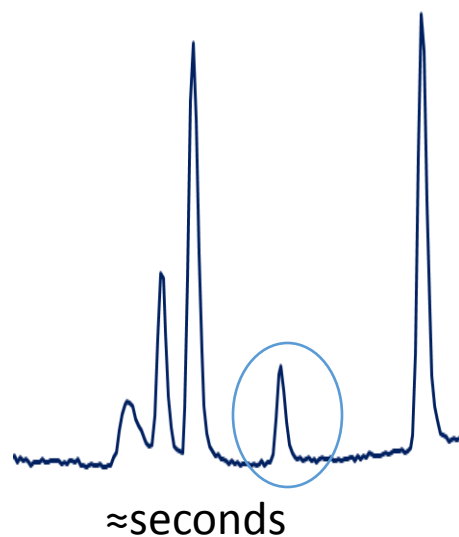
- Additional separation dimension
- Standalone MS or hyphenated with LC
- Several types of ion mobility



Basic operation of ion mobility



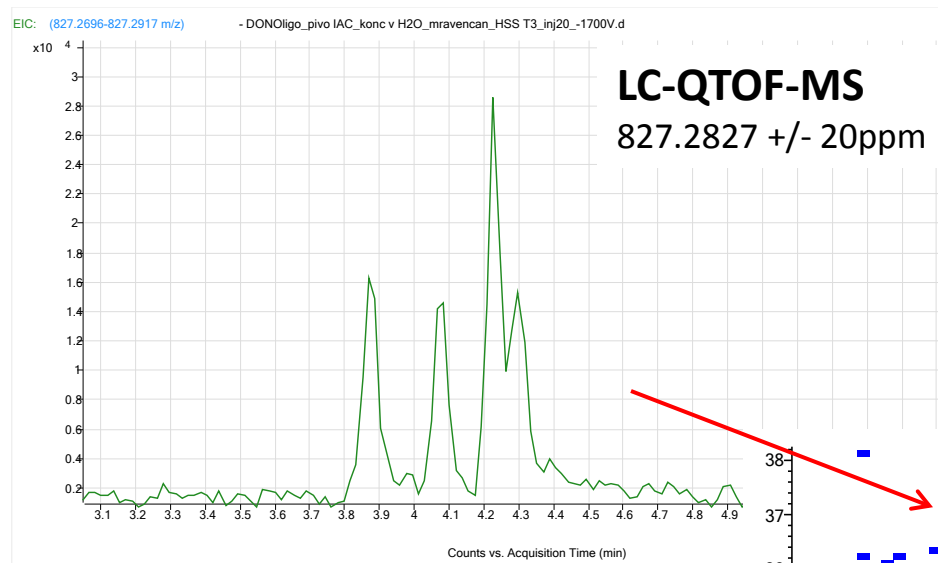
Adding Ion Mobility Spectrometry in LC/MS



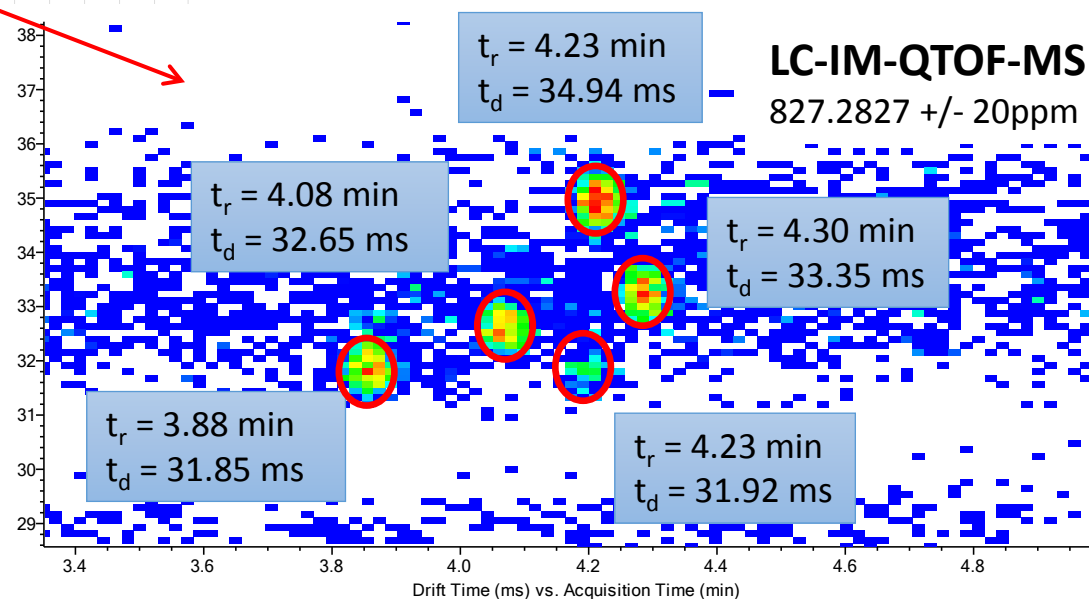
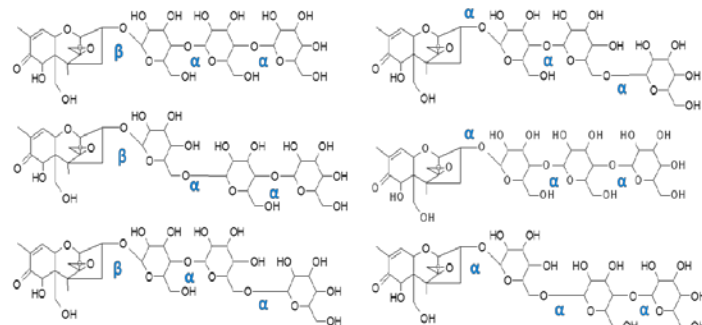
IMS fits between LC and TOF MS on the separation time scale!

IM separation of masked mycotoxins

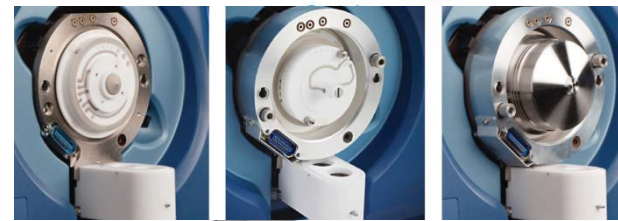
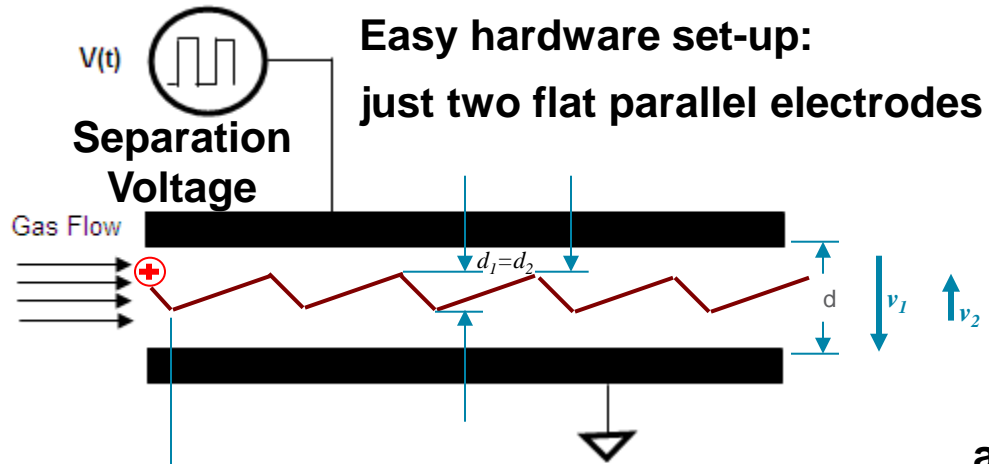
DON-3-triGlc [M+HCOO]⁻



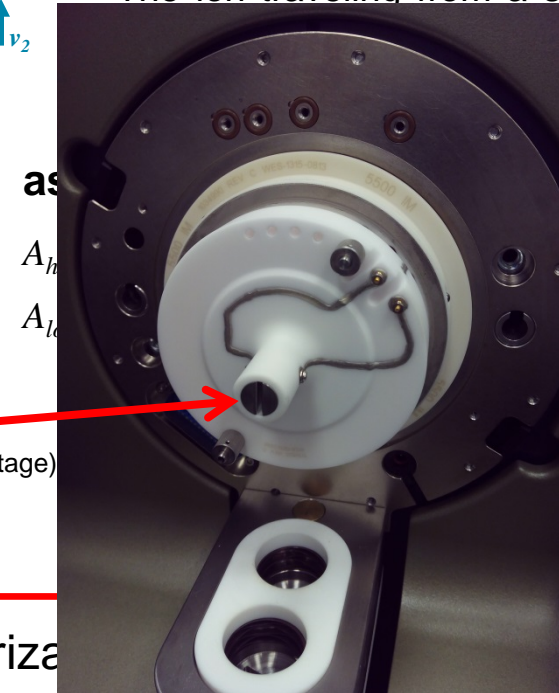
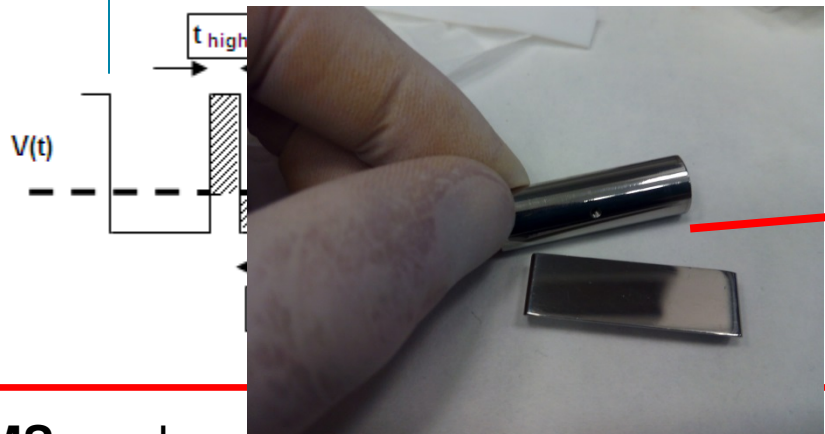
Five isomers found
in IM!



Differential Mobility Spectrometry



The ion traveling from a starting point will travel a distance from



$$v_1 = K \cdot E_1$$

$$v_2 = K \cdot E_2$$

if

DMS can be also understood as the miniaturization

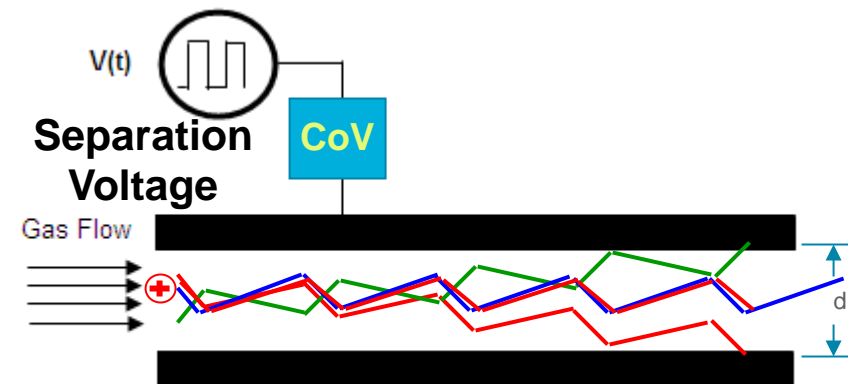
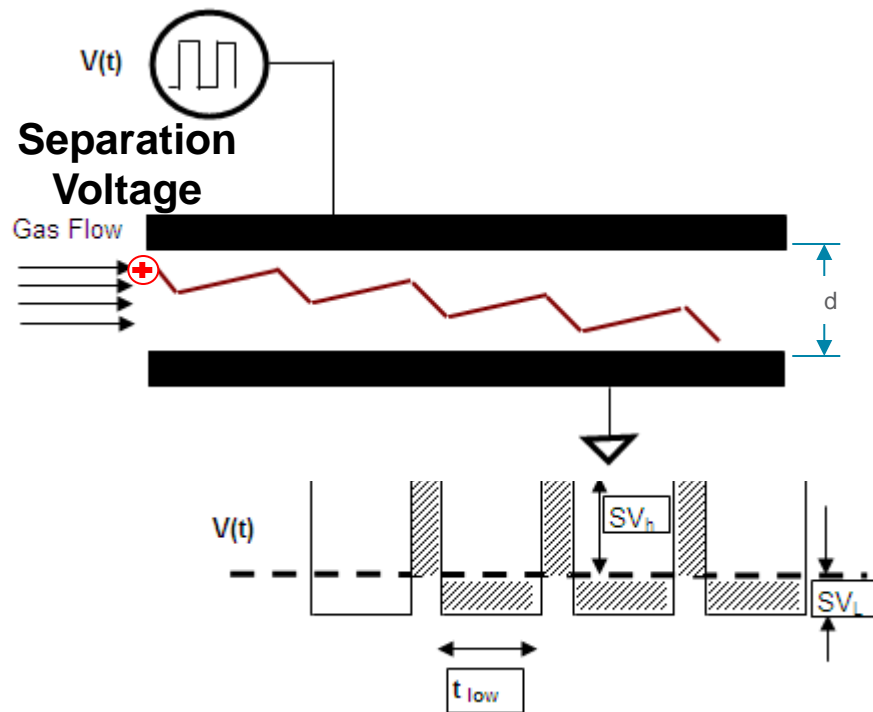
DMS takes advantage of the differences in the mobility of ions in high and low electric fields (**Separation voltage**)

Differential Mobility Spectrometry

However, if the waveform is applied by high SV, the mobility of the ion during application of the peak voltage deviate from its low-field value (dependance of K from E). In this instance, during the higher voltage portion of the waveform, the ion travels at a velocity different than it would absent; this change in mobility:

$$v = K_{(E)} \cdot E$$

The ion traveling from a starting point will therefore not return to exactly this same distance from the electrode after one cycle and, thus, drift towards one of the electrodes.

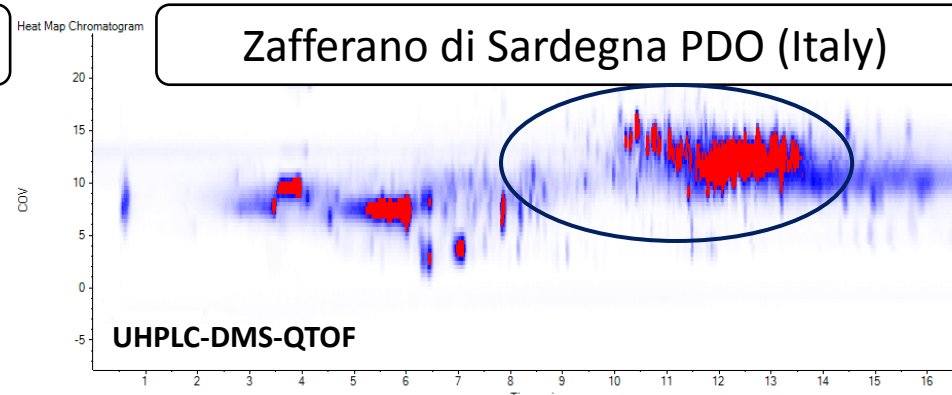
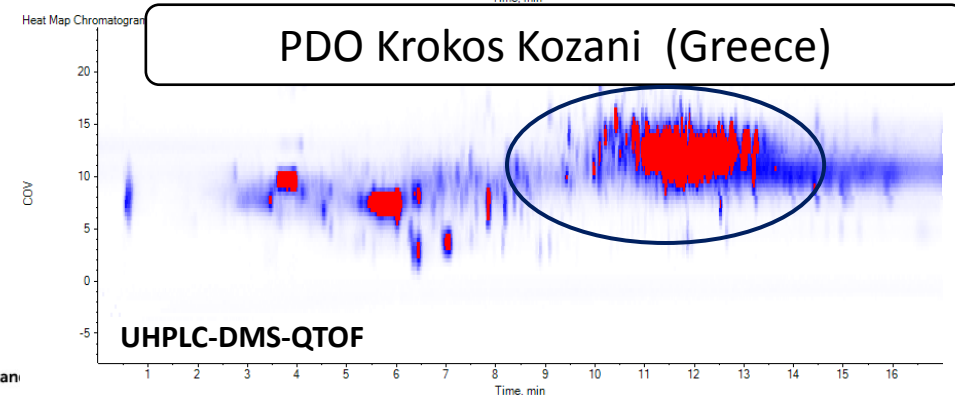
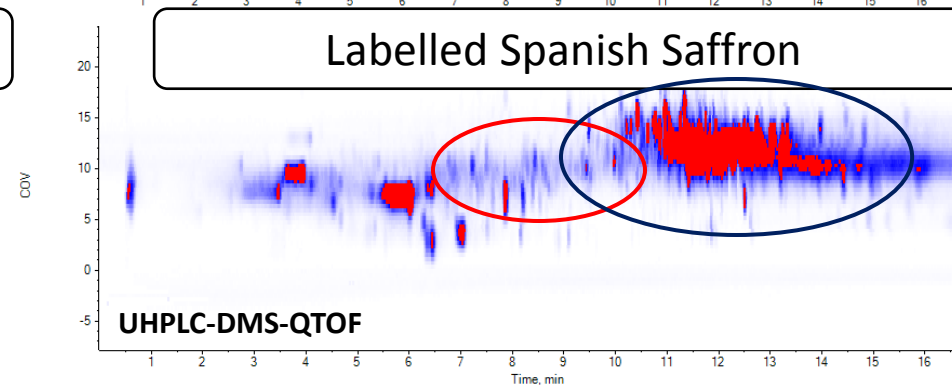
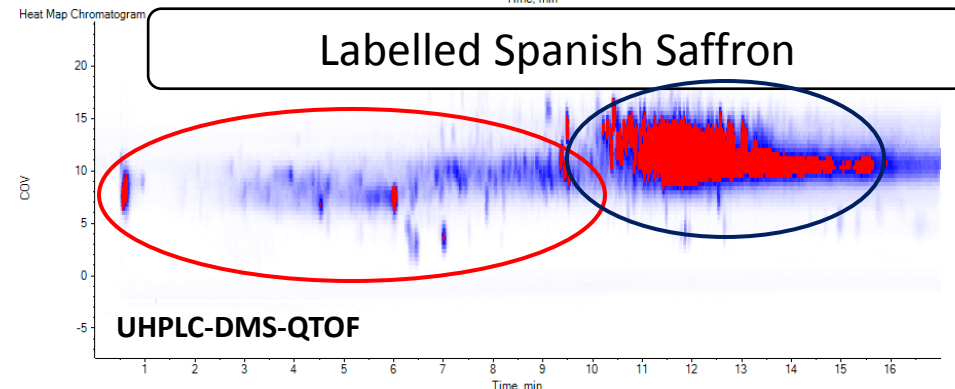
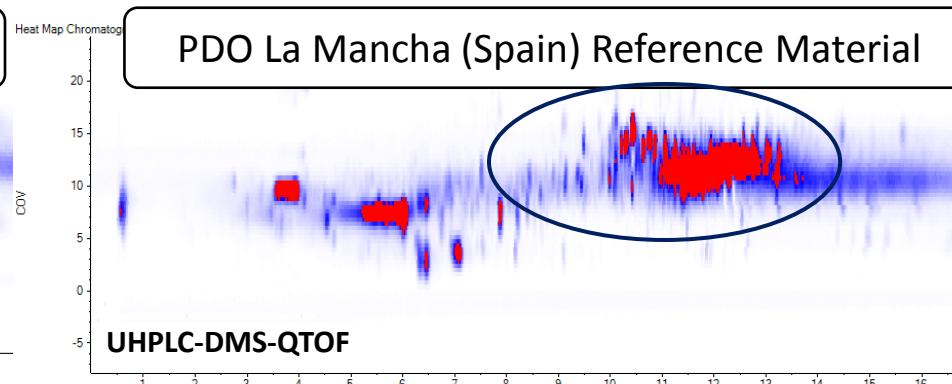
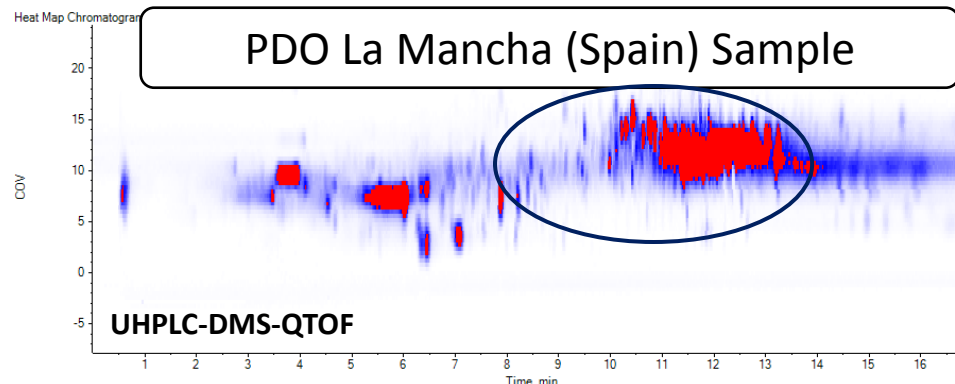


Compensation voltage (CoV):

Restores the trajectory for a given ion to allow them to transmit through the DMS device and enter the mass spectrometer

Heat Map Chromatograms

m/z 100 to 1200
CoV -8 to 24
Run time 17 min





Conclusions

- Wide array of techniques exist
- Domination of separation techniques and mass spectrometry
- Manufacturesrs are routinely assisting in development of methods
- **More exciting instrumental techniques to come!**